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ANSWER: Correct Chapter 8 Reading Question 3 Part A How many phosphorus atoms are on the product side of the following balanced equation? $3\text{MgSO}_4(\text{aq}) + 2\text{K}_3\text{PO}_4(\text{aq}) \rightarrow 3\text{K}_2\text{SO}_4(\text{aq}) + \text{Mg}_3(\text{PO}_4)_2(\text{s})$ Hint 1. The total number of atoms in each formula is obtained by multiplying the coefficients by each subscript in the chemical formula.

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Mastering chemistry Chapter 8. An attractive interaction between a hydrogen atom bonded to a very electronegative atom (O, N, or F). In one molecule and an electron rich regions in a different molecule. (Also could appear within a single molecule) It commonly bonds to unshared electron pairs on another electronegative atom.

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Pearson Chemistry Chapter 8. Covalent Bond. Molecule. Diatomic molecule. Molecular compound. A bond formed by the sharing of electrons between atoms. A neutral group of atoms joined together by covalent bonds. A molecule that contains two atoms. A compound that is composed of molecules.

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